

- 1) The following are useful for making estimates – you should memorize these relations:
  - a. Wavelength of a photon that has 1eV in energy
  - b. Energy (in eV) of a photon with wavelength of 1 $\mu$ m.
  - c. Energy (in eV) for kT at room temperature (300K).
  - d. Show that for light,  $\frac{\Delta\lambda}{\lambda_0} = \frac{\Delta\omega}{\omega_0} = \frac{\Delta\nu}{\nu_0} = \frac{\Delta\sigma}{\sigma_0}$ , where the  $\Delta$ 's correspond to bandwidth, and the denominators correspond to the central carrier frequency (or wavelength...). The symbols, in order are: wavelength, angular frequency, frequency, and wavenumber. In these expressions, you can drop any sign you get in the derivation. These relations are useful to convert spectral bandwidth into different units.
  
- 2) Spectral ranges (see Svelto 1.1 and its solution):
  - a. Look up the transmission or reflection spectral ranges for the following materials:
    - i. Fused silica
    - ii. Crown glass (e.g. BK-7)
    - iii. CaF<sub>2</sub>
    - iv. Silicon
    - v. Gold
    - vi. Silver
    - vii. Aluminum
  - b. Look up the spectral range of sensitivity of the following detectors:
    - i. Human eye
    - ii. Silicon CCD
    - iii. InGaAs
    - iv. Microbolometer
  - c. Look up the spectral ranges of the following categories of lasers:
    - i. Ion lasers (Ar-ion, Kr-ion)
    - ii. HeNe
    - iii. Excimer
    - iv. Ti:sapphire
    - v. Neodymium-doped material (e.g. Nd:YAG, Nd:glass...)
    - vi. Erbium-doped material (Er: )
    - vii. CO<sub>2</sub>
    - viii. Semiconductor diode lasers (not LED's)

- 3) Suppose we have a two-level quantum system, with an energy difference  $\Delta E$  separating the levels. The two levels have the same degeneracy. Fill in the following table for the thermal equilibrium population ratio  $N_2/N_1$  for the following combinations of temperatures  $T$  and energy differences  $\Delta E$ .

	T = 100K	T = 300K	T = 1000K
$\Delta E = 0.0001$ eV			
$\Delta E = 0.05$ eV			
$\Delta E = 3$ eV			

The lowest energy difference is characteristic of rotational transitions in molecules, the next corresponds to molecular vibrational transitions, and the highest energy difference is of the order of magnitude of electronic transitions in atoms and molecules.

- 4) For a plane wave of intensity  $100 \text{ W/m}^2$  ( $=10 \text{ mW/cm}^2$ ),
- Calculate the electric field strength and the energy density in SI units, where  $I = \epsilon_0 c n \langle |E|^2 \rangle$ , where the angular brackets denote a time (cycle average). Assume the wave is in vacuum ( $n=1$ ).
  - Calculate the corresponding photon flux for a monochromatic beam with wavelengths  $500\text{nm}$  and  $100\mu\text{m}$ .
- 5) The intensity of sunlight at the earth's surface is  $1 \text{ kW/m}^2$ . Calculate the intensity of an image of the sun on the retina assuming:
- The eye's pupil diameter is  $2\text{mm}$  (bright-adapted).
  - The focal length of the eye is  $22.5\text{mm}$ .
  - The sun subtends an angle of  $0.5^\circ$ .

*Hint:* first calculate the size of the image of the sun on the retina by using similar triangles for rays passing from both edges of the sun through the center of the lens (undeviated), then to the retina at one focal length from the lens.

Compare this with the intensity of a  $1\text{mW}$  HeNe laser ( $\lambda = 632.8\text{nm}$ ) at the retina. The beam is Gaussian in profile and enters the eye with a  $1/e^2$  diameter of  $2\text{mm}$ .

*Hint:* the  $1/e^2$  radius of a Gaussian beam focused by a lens of focal length  $f$  is

$$w = \frac{2}{\pi} \lambda \frac{f}{D},$$

where  $D$  is the  $1/e^2$  diameter of the input beam. It is expressed in this way

because  $f/D$  is the effective  $F$ -number of the focus.

- 6) Svelto 2.1 (solution is in book). Suppose the bandwidth of a HeNe laser is  $1.5 \text{ GHz}$ . Calculate the bandwidth  $\Delta\lambda$  in wavelength space, then calculate the number of modes in the box in this case.
- 7) Svelto 2.3. Use Mathematica to numerically solve for the actual peak wavelength.